Antibacterial Effects of Fluoride in *Streptococcus mutans* Growth in Vitro

Yura Pradiptama¹, Marijam Purwanta²*, Harianto Notopuro²

¹Faculty of Medicine Universitas Airlangga, Surabaya, Indonesia
²Department of Microbiology, Faculty of Medicine Universitas Airlangga, Surabaya, Indonesia - Dr. Soetomo General Hospital Surabaya, Indonesia

**Article history:**
Received 14 May 2019
Received in revised form 10 June 2019
Accepted 25 June 2019

**Keywords:**
*Streptococcus mutans*, Sodium fluoride, Dilution test, MBC, MIC.

**ABSTRACT**

**Introduction:** *Streptococcus mutans* is a gram-positive coccus commonly found in the human oral cavity and is a pathogen of dental caries. *S. mutans* known to form biofilm in infective endocarditis heart-valve. Fluoride usage known to reduce the risk of dental caries. This study aims to analyze how usage of fluoride in inhibiting *S. mutans* growth.

**Methods:** This study was an experimental study. Antibacterial activity test was performed to evaluate the minimum inhibitory concentration (MIC) using dilution method of sodium fluoride. The minimum bactericidal concentration (MBC) was determined by culturing from the previous dilution test into Chocolate Agar Plate.

**Results:** MIC for sodium fluoride is 4,8 mg/ml and the MBC for sodium fluoride to *S. mutans* is 4,8 mg/ml. We found *S. mutans* growth in higher concentration than 19,2 mg/ml.

**Conclusion:** *S. mutans* was inhibited in dilution test. Growth of the bacteria in higher concentration of sodium fluoride is explained with quasi-irreversible inhibitor effects of fluorida.

© 2018 Biomolecular and Health Science Journal. All rights reserved

---

**Introduction**

*Streptococcus* spp. is the most common bacteria in mouth. This gram positive bacteria grows in facultative anaerobic environment. The most common species that we can found in our mouth is *Streptococcus mutans*.1 Bowden in 1996 found that plaque mainly composed from bacteria which lives in our mouth.2 *S. mutans* believed to be the primary factors this process. It is said that *S. mutans* was the main reason in initiating dental caries by forming biofilm in dental lesion.3 *S. mutans* metabolized sucrose more efficient than any other commensal bacteria and it have a regulation system to stimulates carbohydrates conversion into acidic substance which help them to excel in making acidic environment in our mouth.4 On the other hand, this nature helps the development of plaque in our teeth by using the lactic acid the bacteria metabolized.5 It is also stated that cariogenic nature from *S. mutans* makes this bacteria an opportunistic infection. *S. mutans* also found in the biofilm from infective endocardidive heart valve.6 There are many ways for us in order to maintain our oral hygiene. Brushing our teeth is the most common and frequently used by us nowadays.7 Fluoride was the most common substance use in toothpaste. Sodium fluoride (NaF), tin (II) fluoride (SnF₂), and sodium monofluorophosphate (Na₃PO₃F) are the most common forms of fluoride use in toothpaste to prevent dental caries. Sodium fluoride effectively suppressed oral bacteria growth by inhibiting enzyme which have important role in glycolytic cycle. Enolase enzym which converting 2-phosphogycerate into phosphoenolpyruvate (PEP) is really important in metabolic process which keep *S. mutans* alive. The decrease of phosphoenolpyruvate will make the growth of this bacteria stutter.8 Because of the widely use of fluoride, we want to see its antibacterial effects to *S. mutans* growths in vitro for further evaluation usage of fluoride in oral hygiene.

**Methods**

This research was a laboratory experimental research conducted in Microbiology Laboratory at Faculty of Medicine, Airlangga University. Sodium fluoride was obtained from a chemical store in Yogyakarta. The *S. mutans* obtained from Microbiology Laboratory of Faculty of Medicine Airlangga University.

This research used 8 concentrations with 2 control tubes. 0,5 ml *S. mutans* (0,5 McFarland) in Mueller-Hinton broth as positive control tubes and sodium fluoride 153,6 mg/ml in Mueller-Hinton broth as negative control. Each tube contain serial dilution of sodium fluoride 153,6 mg/ml, 76,8 mg/ml, 38,4 mg/ml, 19,2 mg/ml, 9,6 mg/ml, 4,8 mg/ml, 2,4 mg/ml, and 1,2 mg/ml with 1 ml *S. mutans* in Mueller-Hinton broth, respectively. All tubes then incubated in 37°C for 24 hours in the incubator with microaerophilic environment (10% CO₂) in candle jar method. Four times replication were conducted for high accuracy. The minimum inhibitory concentration (MIC) was determined by visually observing...
the lowest sodium fluoride concentration with no turbidity. For each tube that had a clear scheme, we will culture into Chocolate Agar Plate and incubated for another 24 hours in candle jar. Every Chocolate Agar Plate will be observed to see if there was any growth of *S. mutans* from each culture that we made from the tubes to determine minimum bactericidal concentration (MBC).

**Results**

These are the results for the dilution test to determine the minimum inhibitory concentration (MIC) for sodium fluoride to *S. mutans*. The MIC observed visually by comparing each tube with the positive control tube. We found that bacterial growth was still found in tubes with 153.6 mg/ml, 76.8 mg/ml, 38.4 mg/ml, 2.4 mg/ml, and 1.2 mg/ml concentration had the same turbidity as the positive control tubes. Then we plated *S. mutans* from each concentration into Chocolate Agar Plate to determine MBC. (Table 1) (Figure 1).

Minimum bactericidal concentration (MBC) was visually observed from the culture to determine if there was any growth of bacteria. It was found that 153.6 mg/ml, 76.8 mg/ml, 38.4 mg/ml, 2.4 mg/ml, and 1.2 mg/ml had *S. mutans* growth, whereas 19.2 mg/ml, 9.6 mg/ml, and 4.8 mg/ml could not be found any bacteria growth. We had constant results from the four-time replication. The minimum inhibitory concentration (MIC) for sodium fluoride was 4.8 mg/ml and the minimum bactericidal concentration (MBC) for sodium fluoride was 4.8 mg/ml. It is concluded that sodium fluoride has bactericidal effects on certain concentrations. (Table 2) (Figure 2).

### Table 1. Sodium fluoride Dilution Test for MIC

<table>
<thead>
<tr>
<th>Replication No.</th>
<th>Sodium fluoride concentration</th>
<th>Control</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>153.6 mg/ml (T1)</td>
<td>76.8 mg/ml (T2)</td>
</tr>
<tr>
<td>1</td>
<td>X</td>
<td>X</td>
</tr>
<tr>
<td>2</td>
<td>X</td>
<td>X</td>
</tr>
<tr>
<td>3</td>
<td>X</td>
<td>X</td>
</tr>
<tr>
<td>4</td>
<td>X</td>
<td>X</td>
</tr>
</tbody>
</table>

X: Turbid bacteria growth found, O: Clear bacteria growth not found

### Table 2. Dilution Test Cultured into Chocolate Agar Plate

<table>
<thead>
<tr>
<th>Replication No.</th>
<th>Sodium fluoride concentration</th>
<th>Control</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>153.6 mg/ml (T1)</td>
<td>76.8 mg/ml (T2)</td>
</tr>
<tr>
<td>1</td>
<td>+</td>
<td>+</td>
</tr>
<tr>
<td>2</td>
<td>+</td>
<td>+</td>
</tr>
<tr>
<td>3</td>
<td>+</td>
<td>+</td>
</tr>
<tr>
<td>4</td>
<td>+</td>
<td>+</td>
</tr>
</tbody>
</table>

+ : Bacteria growth found, - : Bacteria growth not found

Figure 1. Sodium fluoride dilution test for MIC

Control tubes to compare which tubes were turbid or clear

Figure 2. Dilution Test Cultured into Chocolate Agar Plate. In T1 (153.6 mg/ml), T2 (76.8 mg/ml), T3 (38.4 mg/ml), T7 (2.4 mg/ml), and T8 (1.2 mg/ml) culture there was growth of *S. mutans*. Meanwhile T4 (19.2 mg/ml), T5 (9.6 mg/ml), and T6 (4.8 mg/ml) there were no growth of any bacteria.
Discussion
Sodium fluoride was used to inhibit enzymes involved in glycolysis and works effectively in low pH which leads to inhibiting the metabolism process. Fluoride can directly inhibit the growth of bacteria. In previous studies it was found that mouth bacteria changes with daily usage of fluoride. It happens because fluoride in this term hydrofluoric acid (HF) will release itself to become H+ and F- in bacteria cell’s which leads to accumulation of fluoride in the bacteria cell and will inhibit the metabolism process. The enzyme which inhibited directly by fluoride is Enolase. High sensitivity to fluoride is the reason why enolase be targeted from the entire process of glycolysis process. Fermentation which really needed to make the process. Enolase activation when the enzyme Exocellular glucosyltrans-ferases (GTFs).

Concentration of 37,000 ppm or 37 mg/ml in pH 3.0. While high concentrations of 5,000 ppm or 5 mg/ml may help reduce biofilm formed. In studies from Bowden (1990), the use of high concentrations of fluoride with a range of 3,040 ppm - 5,700 ppm or 3.04 mg/ml - 5.7 mg/ml are needed to kill bacteria in the mouth. Meanwhile Caufield & Wannenmuhler (1982), found that MBC of sodium fluoride on S. mutans at pH 7-8 has a concentration range of 3,500-5,000 ppm or 3.5-5 mg/ml.

Sodium fluoride gel does not show any antibacterial activity in colonies of S. mutans. The concentration of sodium fluoride needed depends on the pH of the media. Lower pH required least concentration of sodium fluoride to have antibacterial effects. Sodium fluoride gel required concentration of 37,000 ppm or 37 mg/ml in pH 3.0. While only 12,000 ppm or 12 mg/ml required at pH 2.5 and 1,100 ppm or 1.1 mg/ml is needed at pH 2.0. Fluoride works by reducing bacterial tolerance to acids which will most effectively use with lower pH.

This research found that sodium fluoride at a concentration of 38.4 mg/ml and higher contained regrowth of S. mutans. This can be with form of adaptation from S. mutans from the use of high concentration fluoride. High fluoride concentration will have a quasi-irreversible nature of the inhibitor on the enzyme enolase.

The nature of this inhibition may be returned by the accumulated level of the 2-phosphoglycerate or any products from the PEP. Methods to eliminate fluoride-inhibited enolase conduct by forcing to accumulate substrate or products from enolase to bring the reversible effects that help S. mutans to stay alive. It is clear that with the high concentration of sodium fluoride, then S. mutans will return to the process of metabolism.

Conflict of Interest
The author stated there is no conflict of interest

References