Chemical Dermal Exposure Risk Assessment in the Water Treatment Plant of Fertilizer Industry

Rizki Rahmawati¹, Mila Tejamaya²

¹Master of Occupational Health and Safety Program, Faculty of Public Health, Universitas Indonesia, Indonesia

²Department of Occupational Health and Safety, Faculty of Public Health, Universitas Indonesia, Indonesia ^{1,2}Building C, 1st Floor Faculty of Public Health, Universitas Indonesia, Depok, West Java 16424, Indonesia

ABSTRACT

Introduction: In water treatment plants (WTP), chemicals play a crucial role. However, some of these chemicals are hazardous. This study aims to conduct a dermal risk assessment in the WTP of an ammonia and urea production facility. **Methods:** The study was performed in August 2023 and assessed dermal exposure risk for four hazardous chemicals: NaOCl (30%), HCl (60%), H2SO4 (98%), and NaOH (48%), utilizing the Tier 2 RISKOFDERM model. Intrinsic toxicity was evaluated using risk phrases and toxicity information. Potential dermal exposure rates (PERBODY and PERHANDS) were determined based on task group and exposure modifier, while actual dermal exposure rates (AERBODY and AERHANDS) were determined based on clothing type and activity time. Health risk was assessed using actual exposure scores and intrinsic toxicity levels, which were categorized into 10 different levels ranging from 1 to 10. **Results:** The risk phrases indicated that four chemicals possessed a high level of intrinsic toxicity in terms of local effect but no systemic effect. PERBODY and PERHANDS were high (NaOCl, HCl) and low (H2SO4, NaOH). The actual exposure scores were determined to be 1 (high) for NaOCl and HCl, 0.01 (low) for H2SO4, and 0.03 (medium) for NaOH. Health risk values were 8 for NaOCl and HCl, 5 for H2SO4, and 6 for NaOH. **Conclusion:** Health risks in NaOCl and HCl were assigned action priority (AP) 1, followed by NaOH at AP-2, and H2SO4 at AP-3. The study recommends the implementation of control measures encompassing engineering solutions, administration, and personal protective equipment.

Keywords: chemical, dermal, risk assessment, water treatment plant

Corresponding Author:

Mila Tejamaya Email: tejamaya@ui.ac.id, Telephone: +628111810100

INTRODUCTION

A water treatment plant (WTP) is a utility facility that is designed to clean and treat water that will be used in the process. In the water treatment plant, chemicals play a crucial role as biocides and disinfectants, coagulants and flocculants, corrosion and scale inhibitors, and pH control. The market for water treatment chemicals has grown from the early 1990s and was projected to continue expanding through 2020 (Gitis and Hankins, 2018). Many workers across multiple sectors are exposed to chemicals, and the number is expected to increase as the use of chemicals increases. In the United States, skin exposure to chemicals at work is a serious issue. Skin conditions are more prevalent than respiratory disorders, both in terms of incidence and rate. The Bureau of Labor Statistics (BLS) recorded 25,000 recordable skin diseases in 2018, or 2.2 injuries per 10,000 employees, while respiratory ailments accounted for 19,600 illnesses, or 1.7 illnesses per 10,000 employees.

The vast majority of chemicals are easily absorbed through the skin, potentially increasing the amount of the chemicals inhaled from the air and leading to additional health implications. Numerous studies show that workers may absorb toxins through their skin without realizing it (Occupational Safety and Health Administration, 2018). According to data from the BPJS Ketenagakerjaan program on occupational diseases and accidents between 2019 and 2021, the chemical and basic industry sectors are among the top five business sectors with the highest frequency of work accident cases (12.1%).

Cite this as: Rahmawati, R. and Tejamaya, M. (2024) 'Chemical Dermal Exposure Risk Assessment in the Water Treatment Plant of Fertilizer Industry', The Indonesian Journal of Occupational Safety and Health, 13(2), pp. 241-251.

©2024 IJOSH All right reserved.. Open access under CC BY NC–SA license doi:10.20473/ijosh.v13i2.2024.241-251. Received January 17, 2024; 1st revision April 18, 2024; 2nd revision May 16, 2024; Accepted May 29, 2024; Published: August 2024. Published by Universitas Airlangga.

There were 145, 109, and 101 work-related chemical exposure accidents. Indonesia's occupational diseases data for 2019–2022 showed 48, 81, and 6 cases, respectively (Kementerian Ketenagakerjaan RI, 2022).

A health risk assessment can be carried out using several methods. Dermal exposure analysis uses Tier 1 tools such as ECETOCC TRA, MEASE, and EMKG-EXPO-TOOL for basic screening. These tools quickly distinguish between dangerous and non-dangerous situations, providing a conservative estimate of exposure based on a few exposure determinants. They are designed to provide a higher estimate than workplace measurements. Stoffenmanager, Advanced REACH Tool (ART), and Risk Assessment of Occupational Dermal Exposure to Chemicals (RISKOFDERM) are examples of higher-tier tools that could provide more advanced and accurate exposure estimations (Schlueter and Tischer, 2020). The use of risk assessment methods depends on the research objectives.

The Dutch Institute TNO led the European Research Project RISKOFDERM, which involved 15 institutes across 10 member states comprising Germany, Spain, France, Italy, the Netherlands, Austria, Portugal, Finland, Sweden, and the United Kingdom. A toolset for evaluating and controlling health risks from skin exposure is created by RISKOFDERM. This toolkit has an algorithm that evaluates the risk by fusing a hazard score for the substance based on general toxicity and the exposure estimate from the RISKOFDERM model; however, it ignores dermal permeability (Oppl et al., 2003). RISKOFDERM is one of the five models recommended for human risk assessment since it can be applied at virtually any step of manufacturing (Franken et al., 2020). PT X is a subsidiary of a state-owned enterprise (BUMN) that operates in the fertilizer and chemical industry. To meet the need for nitrogen-based fertilizer, in 2018 PT X built a new production unit, namely the Ammonia, Urea, ZA Factory. Based on PT X internal report, the largest use of auxiliary chemicals is in the water treatment plant. Of the six chemicals used, five chemicals have a route of entry through the skin. Notably, sodium hypochlorite (NaOCl), HCI, sulfuric acid (H2SO4), and sodium hydroxide (NaOH) are classified as H Code H314 (skin corrosion) category I B and H318 (eye damage category 1).

The risk of accidents at the WTP is related to chemicals that have a medium to high level of risk (Falakh and Setiani, 2018). Between 2020 and 2023, there were 13 chemical-related accidents, with outcomes ranging from first aid injuries (FAI) to medical treatment injuries (MTI) in PT X. The procedure at WTP involves the use of numerous chemicals, as well as human contact. However, a health risk assessment of chemical exposure to determine the level of health risk (Risk Rating/RR), which is crucial for company management, has never been undertaken.

METHODS

This research was conducted in August 2023 using descriptive research with a semiquantitative approach. Data collection methods included secondary data, field observations, and semiquantitative dermal risk assessment. The RISKOFDERM model, with the variables Dermal Exposure Operation (DEO) units, Potential Dermal Exposure Rate, Actual Dermal Exposure Rate, Actual Exposure Score, and Control selected for the dermal risk study of chemicals in this research.

Hazard information was obtained from risk phrases according to the European Dangerous Substances Directive used in the RISKOFDERM toolkit. Risk phrases are ranked into an intrinsic toxicity (IT) score, which then equals the hazard score (Table 1). Dermal exposure is defined as Dermal Exposure Operation (DEO) units (Figure 1) because it is believed that, in its appropriate dimensions, it can be extrapolated from one component to another when based on a particular activity. DEO aims to group dermal exposure situations that have a similar relationship between exposure and exposure level. Modifying factors (substance-related, workplace-related, and controlrelated) also have an impact on the DEO default value; however, their effects differ depending on the exposure route (direct contact, surface contact, or exposure by deposition) (Van Hemmen, 2005).

 Table 1. Score for Intrinsic Toxicity for Local Health
 Effects

R-Phrase	Intrinsic Toxicity (IT) Score
None of those below	Low
R 66, R 38	Moderate
R 34, pH<2 or pH≥11.5	High
R 35, R 43	Very High
R 45	Extreme

Source: Oppl et al. (2003)

Potential exposure occurs when a chemical reaches the exterior of the body. RISKOFDERM has determined the Default Potential Exposure Rate (DPEBODY and DPEHANDS) in units of mg cm-2 h-1 based on the DEO unit. Potential Dermal Exposure Rate (PER) is the product of the default potential exposure rate and the overall modifier (Equation 1). If the exposed body part is unprotected, then the actual exposure (AER) is assumed to be equal to the potential exposure (PER). However, if protected by clothing or protective equipment (such as gloves, aprons, and helmets), then the amount of protection depends on the percentage of coverage, the thickness of the clothing, and the physical state of the chemicals encountered (dust or liquid) (Oppl et al., 2003).

Actual Dermal Exposure Rate (AER) is the multiplication of PER with a modifying factor, namely 0.5 (light clothing) or 0.1 (thick clothing)

1 Handling of objects
Filling
Collecting
Maintenance and servicing
Loading
Mixing/diluting
2 Manual dispersion of substance
Wiping
3 Dispersion of substance with hand-held tool
Pouring
Spreading with comb
Rolling
Brushing
4 Spray dispersion of substance
Spraying
5 Immersion
Immersing of objects (electroplating)
6 Mechanical treatment (of solid objects)
Machining
Grinding
Sawing

Figure 1. Dermal Exposure Operation (DEO) Units Based on RISKOFDERM Models Source: Van Hemmen (2005)

 $PER_{BODY} = \begin{array}{l} default \text{ potential exposure rate (body)} \\ \times \text{ overall modifiers} & (1a) \end{array}$ $PER_{HANDS} = default \text{ potential exposure rate} \\ (hands) \times \text{ overall modifiers} & (1b) \end{array}$

Equation 1. Potential Dermal Exposure Rate Source: Oppl *et al.* (2003)

(Equation 2). However, RISKOFDERM does not consider the use of protective clothing or gloves because they may not provide adequate protection. In addition, it depends on how they are put on and taken off. Thus, AERHANDS in Equation 2 is the same as PERHANDS.

The actual peak exposure level score (AERPEAKscore) was assigned to the exposure level for the hand, which was determined to be the best estimate of the peak exposure level (AERPEAK). The actual exposure dose (AED) is the product of the AERHANDS score and Activity Time (AT) (Equation 4). The exposed body area (EBA) score is obtained based on information in the field regarding which parts of the body are exposed. Actual Exposure (AE) score is the product of AED score and EBA score (Equation 5).

The health risk score is derived by combining the Hazard score matrix from Intrinsic Toxicity (Table 1) with the Actual Exposure (AE) score (Table 2), resulting in a scale of 10 levels of health risk. The scale ranges from 1, indicating the lowest health risk, to 10, representing the highest risk value (Table 3).

Dermal Exposure Operation (DEO) Units

Sodium hypochlorite and HCI handling activities at the WTP of PT X consisted of manually

 $AER_{BODY} = potential exposure rate (body)$ $\times clothing reduction factor (2a)$ $AER_{HANDS} = potential exposure rate (hands)$ $\times clothing reduction factor (2b)$

Equation 2. Actual Dermal Exposure Rate Source: Oppl *et al.* (2003)

 AER_{PEAK} score = value of actual exposure rate of the hands (see eq. 2b) (3)

Equation 3. Actual Dermal Exposure Rate Peak Source: Oppl *et al.* (2003)

 AED_{PEAK} score = AER_{PEAK} score × AT score (4)

Equation 4. Actual Exposure Dose Source: Oppl *et al.* (2003)

 AE_{PEAK} score = AED_{PEAK} score × EBA score (5)

Equation 5. Actual Exposure Score Source: Oppl *et al.* (2003)

pouring the chemical from pails into the process tank, while sulfuric acid (H_2SO_4) was used in a closed system, ensuring that it was only used when maintenance activities were required. Sodium hydroxide (NaOH) was transported using an isotank, and there was a potential for contact when unloading NaOH from the isotank to the process line. As seen in Figure 1, the four activities are based on DEO Unit 1, namely the handling of contaminated liquid objects with the type of contact surface contact. Contribution Factors (CFB) for DEO Unit 1, as determined by the RISKOFDERM toolkit, are 100% (body and hands). The Default Potential Exposure Rate (DPE) values are 0.2 (high) for the body and 0.656 (high) for the hands.

Potential Dermal Exposure Rate

Potential Dermal Exposure Rate (PER) is a qualitative score obtained based on task group and exposure modifier (substance, workplace, or control). Based on observations in the field with reference to the RISKOFDERM toolkit, the exposure modifiers for NaOC1 and HC1 are 'like water', unrestricted workspace, fully manual, and natural

Table 2. Peak Actual Exposure (AE) Score LocalHealth Effects

AEDPEAK score x EBA score	Actual Exposure AEPEAK scores
0.002 or less	Negligible
>0.002-0.02	Low
>0.02-0.2	Moderate
>0.2-2	High
>2-20	Very high
>20	Extreme

Source: Oppl et al. (2003)

 Table 3. Health Risk Score for Local Health Effects

Actual	Hazard Score (Local)					
Exposure Score (Local)	Low (No Risk)	Moderate	High	Very High	Extreme	
Negligible	1	1	2	5	8	
Low	1	2	5	5	10	
Moderate	2	3	6	6	10	
High	2	4	6	8	10	
Very High	3	7	7	9	10	
Extreme	7	9	9	9	10	

Source: Oppl et al. (2003)

ventilation. Meanwhile, the exposure modifiers for H2SO4 are touch dry/small areas of contamination, unrestricted workspace, fully automated, and natural ventilation. For NaOH, the exposure modifiers are touch dry/small areas of contamination, unrestricted workspace, partially automated partially manual, and natural ventilation. PER is calculated on the body (PERBODY) and on the hands (PERHANDS).

Actual Dermal Exposure Rate

Actual Dermal Exposure Rate (AER) BODY is a qualitative score obtained from the PERBODY value by considering the type of workwear used. Meanwhile, AERHANDS is the same as PERHANDS because the use of hand protection is not taken into account in this assessment.

Actual Exposure Score

Actual Exposure (AE) score is obtained by multiplying the Actual Exposure Dose by the Exposure Body score according to the RISKOFDERM toolkit. Actual Exposure Dose is obtained based on the AER Hands score multiplied by the Activity Time score.

Control

The risk of chemicals to health is displayed by the health score value, which is a combination between the AE score and the Hazard score. In the event that the resulting risk is sufficiently low, no further requirements arise from the risk assessment. However, if not, additional control measures are needed and a new risk assessment is recommended to determine their effectiveness. The RISKOFDERM project group established several risk control efficiency classes following European law (Chemical Agents at Work Directive 98/24/EEC), namely the

 Table 4. Action Priority Determination Based on CHRA DOSH Malaysia

Level of Risk	A d e q u a c y Control	of Action Priority (AP)
High	Inadequate	1
Health Risk Could Not Be Determined	-	
Moderate/ Low	Inadequate	2
High/Moderate/ Low	Adequate	3

Source: Department of Occupational Safety and Health Ministry of Human Resources Malaysia (2018) STOP principle: Substitution, Technical Protection, Organizational Protection, and Personal Protection (Van Hemmen, 2005). The Chemical Health Risk Assessment (CHRA) of the Department of Occupational Safety and Health (DOSH) Malaysia is used to determine action priorities based on the level of risk and adequacy of control (Table 4).

RESULTS

Chemical Health Risk Assessment using the RISKOFDERM model was carried out at the WTP unit. From the results of this analysis, the four chemicals used, sodium hypochlorite, HCI, sulfuric acid (H2SO4), and sodium hydroxide (NaOH), had a health hazard with the risk phrases R34 (causes burns) with an intrinsic toxicity score in the high category and were not identified as having systemic effects.

Table 5 shows the complete computation and analysis results of RISKOFDERM. The potential dermal exposure rate (PERBODY and PERHANDS) was in the high category for NaOCl and HCl and in the low category for H2SO4 and NaOH. Meanwhile, the actual dermal exposure rate of AERBODY was 0.1 times that of PERBODY, while AERHANDS was the same as PERBODY. The Actual Exposure score was high for two chemicals (sodium hypochlorite and HCI), medium for NaOH, and low for H₂SO₄. The results of the health risk score analysis was 8 for two chemicals, 6 for one chemical, and 5 for one chemical, on a scale of 10. The results are presented in Table 5.

DISCUSSION

NaOCl and HCl

Sodium hypochlorite is generally used in solution form, while its solid form is not used for commercial purposes. The solution is a clear liquid, yellowish green in color with a chlorine odor. If it reacts with acid, it can release chlorine gas (Pubchem, 2023c). The benefits of hypochlorite treatment include a simple, quick, and efficient procedure, a high capacity, minimal sludge production, the ability to recycle water, and disinfection by bacteria and viruses (Crini and Lichtfouse, 2019).

Sodium hypochlorite, a known irritant, has also been reported to cause type IV allergic contact dermatitis, depending on exposure time and concentration. Some other reported cases include immediate urticarial rash (with 0.1% sodium hypochlorite solution), edema and progression to erythema lesions, necrosis of subcutaneous tissue layers and third-degree burns (>4% sodium hypochlorite). This case has been reported to arise due to accidents or exposures to sodium hypochlorite both in the work environment and at home (Chung et al., 2022). The potential toxicity of hypochlorite is related to its ability to oxidize and the pH of its solution, causing damage to the skin and mucous membranes due to its corrosive properties. Prolonged or extensive skin exposure can cause skin irritation and damage, as well as dermal hypersensitivity, with the possibility of immediate or delayed skin reactions, especially in high-concentration solutions that can cause serious chemical burns (Slaughter et al., 2019).

Hydrochloric acid is a clear, sharp-smelling solution that is often used in chloride, fertilizer, dye, electroplating, textile and rubber industries. It is corrosive to the eyes, skin, and mucous membranes in acute exposure, and chronic exposure can cause dermatitis. The Environmental Protection Agency (EPA) does not classify this substance as carcinogenic.

More concentrated bleach contains 10-15% sodium hypochlorite, which has a pH of approximately 13 and is corrosive. In contrast, household bleach typically contains around 5% sodium hypochlorite, with a pH of about 11 and can cause irritation (ILO, 2021). In the WTP unit of PT X, the concentration of NaOCl used was 30% and HCl was 60%; therefore, they are corrosive to the dermis. Exposure to hydrochloric acid can cause skin burns, ulcerations, and vision problems as side effects. Repeated exposure may cause dermatitis (Pubchem, 2023a).

The calculation of the Potential Dermal Exposure Rate (PERBODY and PERHANDS) indicated high exposure levels to NaOCl and HCl due to the manual chemical addition activities in the PT X's WTP unit. Pouring liquids from buckets into the reaction tank poses a risk of exposure to body parts, such as hands, forearms, and head (2,800 cm2). According to field operators, the duration of this work was around 15 minutes per shift. The company provided personal protective equipment, specifically rubber gloves and face shields; however, based on the observations in the field, compliance with the use of this equipment was still low. This was evidenced by an incident in September 2023

Variable	NaOCl (30%)	HCl (60%)	H2SO4 (98%)	NaOH (48%)
Hazard Score/ Intrinsic Toxicity (IT) Score	H314, H318 Risk Phrases: R34 (Causes Burns) Local Effect ITL Score: High	H314, H318 Risk Phrases: R34 (Causes Burns) Local Effect ITL Score: High	H314, H318 Risk Phrases: R34 (Causes Burns) Local Effect ITL Score: High	H314 Risk Phrases: R34 (Causes Burns) Local Effect ITL Score: High
Dermal Exposure Operation (DEO)	Pouring the chemical from the pail into the process tank DEO = 1 (Handling of Contaminated Objects (Liquid) Type of contact = Surface Contact (SC)	Pouring the chemical from the pail into the process tank DEO = 1 (Handling of Contaminated Objects (Liquid) Type of contact = Surface Contact (SC)	Maintenance of line H2SO4 DEO = 1 (Handling of Contaminated Objects (Liquid) Type of contact = Surface Contact (SC)	Connect line NaoH from isotank into process line DEO = 1 (Handling of Contaminated Objects (Liquid) Type of contact = Surface Contact (SC)
Contribution Factors (CFB in %)	Body Exposure = CFB_{SC} 100% Hand _{s Exposure =} CFH_{SC} 100%	Body _{Exposure = CFBSC} 100% Hands Exposure = CFH _{SC} 100%	Body _{Exposure} = CFBSC 100% Hands Exposure = CFH _{SC} 100%	Body $_{Exposure} = CFBSC$ 100% Hands Exposure = CFH _{SC} 100%
Default Potential Exposure Rate (DPE)	$DPE_{BODY} = 0.2$ (High) $DPE_{HANDS} = 0.656$ (High)	$DPE_{BODY} = 0.2$ (High) $DPE_{HANDS} = 0.656$ (High)	$DPE_{BODY} = 0.2 (High)$ $DPE_{HANDS} = 0.656$ (High)	$DPE_{BODY} = 0.2$ (High) $DPE_{HANDS} = 0.656$ (High)
Exposure Modifier				
Correction Factors for Substance-related Modifiers Integrated Modifying Factor (MFI) $MFI_{BODY} = MF_{SC} x$ CFB_{SC} $MFI_{HANDS} = MF_{SC} x$ CFH_{SC}	Like water (MF _{SC} = 1) MFI _{BODY} = MF _{SC} x CFB _{SC} = 1 x 100% = 1 MFI _{HANDS} = MF _{SC} x CFH _{SC} = 1 x 100% = 1	Like water (MF _{SC} = 1) MFI _{BODY} = MF _{SC} x CFB _{SC} = 1 x 100% = 1 MFI _{HANDS} = MF _{SC} x CFH _{SC} = 1 x 100% = 1	Touch dry/ small areas of contamination (<20%) $(MF_{SC} = 0.1)$ $MFI_{BODY} = MF_{SC} x$ CFB_{SC} = 0.1 x 100% = 0.1 $MFI_{HANDS} = MF_{SC} x$ CFH_{SC} = 0.1 x 100%	Touch dry/ small areas of contamination (<20%) $(MF_{SC} = 0.1)$ $MFI_{BODY} = MF_{SC} x$ CFB_{SC} = 0.1 x 100% = 0.1 $MFI_{HANDS} = MF_{SC} x$ CFH_{SC} = 0.1 x 100%
Correction Factors for Workplace- related Modifiers Integrated Modifying Factor (MFI) $MFI_{BODY} = MF_{SC} x$ CFB_{SC} $MFI_{HANDS} = MF_{SC} x$ CFH_{SC}	Unrestricted workspace $(MF_{SC} = 1)$ $MFI_{BODY} = MF_{SC} x$ CFB_{SC} = 1 x 100% = 1 $MFI_{HANDS} = MF_{SC} x$ CFH_{SC} = 1 x 100% = 1	Unrestricted workspace $(MF_{SC} = 1)$ $MFI_{BODY} = MF_{SC} x$ CFB_{SC} = 1 x 100% = 1 $MFI_{HANDS} = MF_{SC} x$ CFH_{SC} = 1 x 100% = 1	= 0.1 Unrestricted workspace $(MF_{SC} = 1)$ $MFI_{BODY} = MF_{SC} x$ CFB_{SC} = 1 x 100% = 1 $MFI_{HANDS} = MF_{SC} x$ CFH_{SC} = 1 x 100% = 1	= 0.1 Unrestricted workspace $(MF_{SC} = 1)$ $MF_{SC} = 0.1$) $MFI_{BODY} = MF_{SC} x$ CFB_{SC} = 0.1 x 100% = 0.1 $MFI_{HANDS} = MF_{SC} x$ CFH_{SC} = 0.1 x 100% = 0.1

Table 5. Results of Risk Analysis of Dermal Exposure of Four Chemicals (NaOCl, HCl, H2SO4, NaOH) at theWater Treatment Plant Unit of PT X in 2023

Variable	NaOCl (30%)	HCl (60%)	H2SO4 (98%)	NaOH (48%)
Correction Factors for Control-related Modifiers MFI _{BODY} = MF _{SC} x CFB _{SC} MFI _{HANDS} = MF _{SC} x CFH _{SC}	No automation, fully manual ($MF_{SC} = 1$) $MF_{IBODY} = MF_{SC} x$ CFB_{SC} = 1 x 100% = 1 $MFI_{HANDS} = MF_{SC} x$ CFH_{SC} = 1 x 100% Natural Ventilation ($MF_{SC} = 1$) $MFI_{BODY} = MF_{SC} x$ CFB_{SC} = 1 x 100% = 1 $MFI_{HANDS} = MF_{SC} x$ CFH_{SC} = 1 x 100%	No automation, fully manual (MF _{SC} = 1) Natural Ventilation (MF _{SC} = 1) MFI _{BODY} = MF _{SC} x CFB _{SC} = 1 x 100% = 1 MFI _{HANDS} = MF _{SC} x CFH _{SC} = 1 x 100% Natural Ventilation (MF _{SC} = 1) MFI _{BODY} = MF _{SC} x CFB _{SC} = 1 x 100% = 1 MFI _{HANDS} = MF _{SC} x CFH _{SC} = 1 x 100%	Fully automated (MF _{SC} = 0.1) Natural Ventilation (MF _{SC} = 1) MFI _{BODY} = MF _{SC} x CFB _{SC} = 0.1 x 100% = 0.1 MFIH _{ANDS} = MF _{SC} x CFHSC = 0.1 x 100% = 0.1 Natural Ventilation (MF _{SC} = 1) MFI _{BODY} = MF _{SC} x CFB _{SC} = 1 x 100% = 1 MFI _{HANDS} = MF _{SC} x CFH _{SC} = 1 x 100%	Partially Automated partially manual $(MF_{SC} = 0.3)$ Natural Ventilation $(MF_{SC} = 1)$ MFI _{BODY} = MF _{SC} x CFB _{SC} = 0.3 x 100% = 0.3 MFI _{HANDS} = MF _{SC} x CFH _{SC} = 0.3 x 100% = 0.3 Natural Ventilation $(MF_{SC} = 1)$ MFI _{BODY} = MF _{SC} x CFB _{SC} = 1 x 100% = 1 MFI _{HANDS} = MF _{SC} x
Modifying Factor per Group (MFG)	Modifying Factor per Group (MFG)	Modifying Factor per Group (MFG)	Modifying Factor per Group (MFG)	Modifying Factor per Group (MFG)
MFG(a) _{BODY} = MFI _{BODY} - MODIFIER 1 x MFI _{BODY} - MODIFIER 2	$\begin{split} MFG(a)_{BODY} &= 1\\ MFG(a)_{HANDS} &= 1\\ MFG(b)_{BODY} &= 1\\ MFG(b)_{HANDS} &= 1 \end{split}$	$\begin{split} MFG(a)_{BODY} &= 1\\ MFG(a)_{HANDS} &= 1\\ \\ MFG(b)_{BODY} &= 1\\ MFG(b)_{HANDS} &= 1 \end{split}$	$\begin{split} MFG(a)_{BODY} &= 0.1\\ MFG(a)_{HANDS} &= 0.1\\ MFG(b)_{BODY} &= 1\\ MFG(b)_{HANDS} &= 1 \end{split}$	$\begin{split} MFG(a)_{BODY} &= 0.1\\ MFG(a)_{HANDS} &= 0.1\\ MFG(b)_{BODY} &= 1\\ MFG(b)_{HANDS} &= 1 \end{split}$
x MFI _{BODY} - MODIFIER 3 x MFG(a) _{HANDS} = MFI _{HANDS} -	$MFG(c)_{BODY} = 1 x 1 = 1$ MFG(c) _{HANDS} = 1 x 1 = 1	$MFG(c)_{BODY} = 1 x 1 = 1$ MFG(c) _{HANDS} = 1 x 1 = 1	$MFG(c)_{BODY} = 0.1 x$ $1 = 0.1$ $MFG(c)_{HANDS} = 0.1 x$ $1 = 0.1$	$MFG(c)_{BODY} = 0.3 x$ 1 = 0.3 $MFG(c)_{HANDS} = 0.3 x$ 1 = 0.3
MODIFIER 1 x MFI _{HANDS} - MODIFIER 2 x MFI _{HANDS} - MODIFIER 3 x	$MFT_{BODY} = 1 \times 1 \times 1 = 1$ $MFT_{HANDS} = 1 \times 1 \times 1$ $= 1$	$MFT_{BODY} = 1 x 1 x 1$ $= 1$ $MFT_{HANDS} = 1 x 1 x$ $1 = 1$	$ \begin{array}{l} MFT_{BODY} = 0.1 \ x \ 1 \ x \\ 0.1 = 0.01 \\ MFT_{HANDS} = 0.1 \ x \ 1 \ x \\ 0.1 = 0.01 \end{array} $	$MFT_{BODY} = 0.1 x 1 x 0.3 = 0.03 MFT_{HANDS} = 0.1 x 1 x 0.3 = 0.03$
Total Modifying Factor (MFT _{BODY}) = MFG(a) _{BODY} x MFG(b) _{BODY} x MFG(c) _{BODY}				
$(MFT_{HANDS}) = MFG(a)_{BODY} x$ MFG(b)_{BODY} x MFG(c)_{BODY}				

Advanced Table 5. Results of Risk Analysis of Dermal Exposure of Four Chemicals (NaOCl, HCl, H₂SO₄, NaOH)at the Water Treatment Plant Unit of PT X in 2023

Advanced Table 5. Results of Risk Analysis of Dermal Exposure of Four Chemicals (NaOCl, HCl, H₂SO₄, NaOH)at the Water Treatment Plant Unit of PT X in 2023

Variable	NaOCl (30%)	HCl (60%)	H2SO4 (98%)	NaOH (48%)
Potential Dermal Exposure Rate (mg cm ⁻² h ⁻¹)	$PER_{BODY} = 1 \times 0.2$ $= 0.2 \text{ (High)}$	$\frac{\text{PER}_{\text{BODY}} = 1 \text{ x } 0.2}{= 0.2 \text{ (High)}}$	$PER_{BODY} = 0.01 \text{ x } 0.2$ = 0.002 (Low)	$PER_{BODY} = 0.03 \times 0.2$ = 0.006 (Low)
$PER_{BODY} = MFT_{BODY} x$ DPE_{BODY}	$PER_{HANDS} = 1 x$ 0.656 = 0.656 (High)	PER _{HANDS} = 1 x 0.656 = 0.656 (High)	$PER_{HANDS} = 0.01 x$ 0.656 = 0.00656 (Low)	$PER_{HANDS} = 0.03 x$ 0.656 = 0.01968 (Low)
$\begin{array}{l} \text{PER}_{\text{HANDS}} = \text{MFT}_{\text{HANDS}} \ \text{x} \\ \text{DPE}_{\text{HANDS}} \end{array}$	0.000 (111gh)		0.00000 (100)	0.01700 (20%)
$\begin{array}{l} \mbox{Actual Dermal Exposure} \\ \mbox{Rate (mg cm^{-2} h^{-1})} \\ \mbox{Table 15 Transformation of Potential Into Actual Dermal Exposure Rate} \\ \hline \mbox{Tick } \mbox{Type of clothing } & \mbox{AER}_{mach} \\ \mbox{Bex } & \mbox{worn } & \mbox{Migners h} \\ \mbox{Migners h} & \mbox{AER}_{mach} \\ \mbox{Migners h} & \mbox{Migners h} \\ \mbox{Migners h} & $	Heavy Work Clothing AER _{BODY} = $0.1 \ge 0.2$ = $0.02 \ge \text{mg cm}^{-2} \text{h}^{-1}$	Heavy Work Clothing AER _{BODY} = $0.1 \ge 0.2$ = $0.02 \text{ mg cm}^{-2} \text{ h}^{-1}$	Heavy Work Clothing AER _{BODY} = $0.1 \ge 0.002$ = $0.0002 \text{ mg cm}^{-2} \text{ h}^{-1}$	Heavy Work Clothing AER _{BODY} = 0.1 x 0.006 = $0.0006 \text{ mg cm}^{-2} \text{ h}^{-1}$
Summer citability (e.g. T-bibts, shorts) # 0.5 x PERecor # PERevects Heavy sork citability # 0.1 x PERecor # PERevects PER Proteintal Demain Exposure Rate AER Actual Demain Exposure Rate AER Actual Demain Exposure Rate The value without units of the Actual Exposure Rate of the hands from Table 15 is assigned the AER scores	$AER_{HANDS} = PERHANDS$ $= 0.656 \text{ mg cm}^{-2} \text{ h}^{-1}$	$AER_{HANDS} = PERHANDS$ $= 0.656 \text{ mg cm}^2 \text{ h}^{-1}$	$AER_{HANDS} = PERHANDS$ $= 0.00656 \text{ mg cm}^{-2} \text{ h}^{-1}$	$AER_{HANDS} =$ PERHANDS = 0.01968 mg cm ⁻² h ⁻¹
Exposed Body Parts	Lower Arms = 1,400 cm^2 Head = 1,400 cm^2 Total = 2,800 cm^2 EBA Score = 1	Lower Arms = 1,400 cm^2 Head = 1,400 cm^2 Total = 2,800 cm^2 EBA Score = 1	Small Area <10 cm ² EBA Score = 0.1	One Hand or Less = 900 cm ² EBA Score = 0.3
Activity Time (AT)	Duration = 0.1 - < 0.5 h/day AT Score = 3	Duration = 0.1 - < 0.5 h/day AT Score = 3	Duration = <0.1 h/day AT Score = 1	Duration = 0.1 - < 0.5 h/day AT Score = 3
Actual Exposure Dose (AED) AED = AER (HANDS) x AT Score	$AED = 0.656 \times 3$ $= 1.968 \rightarrow AED$ Score = 1	$AED = 0.656 \times 3$ $= 1.968 \rightarrow AED \text{ Score}$ $= 1$	$AED = 0.00656 \times 1$ = 0.00656 $\rightarrow AED$ Score = 0.1	$AED = 0.01968 \times 3$ = 0.05904 $\rightarrow AED$ Score = 0.3
Actual Exposure (AE) Score AE = AED Score x EBA Score	AE Score = 1 x 1 = 1 (High)	AE Score = 1 x 1 = 1 (High)	AE Score = 0.1 x 0.1 = 0.01 (Low)	AE Score = 0.3 x 0.3 = 0.09 (Moderate)
Health Risk Score Matrix AE Score dan Hazard Score	Health Risk Score = 8 (Only exceptionally tolerable, substitute, if any possible)	Health Risk Score = 8 (Only exceptionally tolerable, substitute, if any possible)	Health Risk Score = 5 (Hazard reduction desirable)	Health Risk Score = 6 (Action necessary: mixtures of measures, priority for detailed analyses)

where a worker suffered from skin irritation on his hands after being splashed with this liquid. Moreover, field operators never received proper training or awareness regarding chemical handling, leading to insufficient risk control measures for NaOCl and HCl.

Based on the analysis results, the Health Risk score of NaOCl and HCl was 8 (high), indicating that they were only exceptionally tolerable. Therefore, if possible, it is recommended to substitute and implement additional control measures. Referring to the CHRA DOSH Malaysia action priority determination (Table 4), risks related to exposure to NaOCl and HCl at PT X had action priority 1.

RISKOFDERM formulates risk control through a STOP principle of control: Substitution, Technical Protection, Organizational Protection, and Personal Protection. The hazard control concepts are identical to those outlined in NIOSH's hierarchy of control, which include elimination, substitution, engineering control, administration, and personal protective equipment (NIOSH, 2023). There are several recommendations that can be implemented at the WTP unit of PT X. Firstly, changing from manual chemical pouring to using a pump as an engineering control measure is highly recommended. Additionally, providing training related to chemical handling and its health risks would be beneficial as an administrative control measure. Lastly, it is essential to ensure the suitability and proper use of PPE such as rubber gloves and face shields, while also emphasizing the importance of personal hygiene after contact with chemicals. This is in line with other studies on chemical industry that found a correlation between knowledge, attitudes, and comfort toward PPE compliance (Aslamiah and Kurniawan, 2019; Noviarmi and Prananya, 2023; Saputra and Widowati, 2023).

H2SO4

Sulfuric acid is extremely harmful to the skin, eyes, and mucous membranes. There have been reports of eye injury caused by contact with sulfuric acid from car batteries. The most prevalent injuries are chemical burns to the conjunctiva and cornea, as well as iritis. Dermal burns caused by exposure to sulfuric acid can be fatal (Pubchem, 2024). Exposure to the eyes or skin results in severe burns, the extent of which varies with the potency of the acid. Swallowing may lead to intense irritation of the mouth and stomach. Sulfuric acid accounted for the primary cause (42%) of chemical burns in the Tarapur industrial complex, India, from 2014 to 2015. Half of these sulfuric acid burns involved an acid concentration of 98% (Kulkarni and Jeffery, 2018).

Sulfuric acid (H_2SO_4) used in the utility unit process of PT X originated from its own manufacturing process which was distributed through a closed system using piping, with contact only occurring during maintenance process. The Actual Exposure (AE) score was 0.01 (low) due to the automated addition of H_2SO_4 , which was rarely carried out. The assessment results for the Health Risk score revealed a moderate level of 5 out of 10.

The company provided safety showers, acid hazard signs, and adequate personal protective equipment, such as acid-resistant clothing and gloves. However, field operators lacked awareness or training related to chemical handling. Until 2023, no accidents had occurred due to H_2SO_4 splashing at PT X, indicating sufficient control over this risk. Referring to the action priority determination of CHRA DOSH Malaysia (Table 4), the risks related to H_2SO_4 exposure at PT X was categorized as action priority 3. Recommendations include offering training on chemical handling and health risks, as well as ensuring compliance with the use of PPE when conducting maintenance activities.

NaOH

Sodium hydroxide is extremely caustic to a variety of materials and can cause severe chemical burns to the eyes and skin even at low aqueous concentrations (Riddick, 2020). When applied as a solid or as a 50% solution, sodium hydroxide (NaOH) is extremely corrosive and can burn the eyes, skin, and mucous membranes severely. Application of a 5% solution to the skin resulted in significant necrosis after four hours in rabbit trials. Rapidly irrigating the eyes after using a 1% solution poses no harm. Solutions higher than 30% severely corrode skin (Pubchem, 2023b). Based on CLP Regulation No. 1272/2008 Annex VI for acute exposure with short-term local effects, the concentration limit for NaOH corrosivity is considered to be 2% (ECHA, 2023). Sodium hydroxide (22%) was the second most common chemical involved in chemical burns in the Tarapur industrial complex in India from 2014 to 2015 (Kulkarni and Jeffery, 2018).

Sodium hydroxide (NaOH) used in the WTP of PT X had a concentration of 48%. It was transported by the supplier using an isotank, and unloading was carried out on the connector which was already in line with the process. Possible contact with chemicals may occur during product unloading from the isotank and connection process to the NaOH line.

The NaOH connection process from the isotank to the system was carried out by the driver supplier on a weekly basis. The personal protective equipment provided was rubber gloves and face shields, but compliance with their use needed to be ensured. Training on chemical handling and its health risks had never been conducted; therefore, a training session needs to be scheduled. This suggests that control over this risk was not sufficient. According to a study on chemical handlers, a significant portion of chemical managers and handlers are unaware of personal protective equipment (PPE) regulations, emphasizing the importance of increased education and the development of educational content to ensure compliance and safety when handling hazardous chemicals (Han, 2021).

The Health Risk score assessment yielded a moderate result of 6. Based on the action priority determination of CHRA DOSH Malaysia (Table 4), the risks related to NaOH exposure at PT X were classified as action priority 2, requiring for the need for measures such as providing training on chemical handling and its health risks, as well as ensuring compliance with the use of PPE for isotank drivers during unloading activities. According to Manzoor (2020), emergency management competencies of hazardous materials (HAZMAT) truck drivers include prevention, preparedness, mitigation, response, and recovery.

Prevention involves safe driving tactics and knowledge of HAZMAT properties. Mitigation involves first aid, triage, and spill control techniques. Preparedness involves understanding emergency response plans, PPE, and emergency services. Response involves communication, reporting mechanisms, and cordoning off areas to protect people and assets. HAZMAT truck drivers educated as first responders can greatly contribute to emergency mitigation, particularly in low- and middle-income countries, through initial training, on-the-job training, refresher courses, and frequent exercises and drills.

CONCLUSION

Health risk values were 8 for NaOCl and HCI, 5 for H_2SO_4 , and 6 for NaOH on a scale of 10. Action priority (AP) 1 was assigned to health risks in NaOCl and HCl handling activities, AP-2 was designated for NaOH, and AP-3 for H2SO4. Recommendations for control can be carried out in accordance with the hierarchy of control. Control that can be carried out through engineering control includes the installation of a pump to facilitate the transfer of chemicals from containers to process tanks. This method prevents workers from coming into direct contact with the chemicals when pouring them. It is important for the company to internally reassess the appropriate pump design.

The next control is administrative control, which includes providing training on chemical handling and its health risks, both for internal workers and third-party chemical providers. The training covers emergency management competencies for hazardous materials (HAZMAT), including prevention, preparedness, mitigation, response, and recovery. Another administrative control is ensuring standard operational procedures are carried out properly. In addition, safety data sheets (SDS), signs, safety showers, and eye wash must also be available at the location. The final control hierarchy is the use of PPE. Companies should ensure the suitability and use of PPE, such as coveralls, face shields, rubber gloves, and safety shoes. Workers, both organic and thirdparty employees, must ensure the proper use of PPE during contact with chemicals, as well as personal hygiene after contact with chemicals.

CONFLICT OF INTEREST

The authors declare that there are no substantial competing financial, professional, or personal interests that could have influenced the performance or presentation of this study.

AUTHOR CONTRIBUTOR

RR recognized the problem, developed the assessment, collected data, assessed it, and concluded the assessment results. MT provided guidance on data assessment, reviewed the manuscript, and approved the manuscript.

ACKNOWLEDGMENTS

The main author would like to acknowledge the water treatment plant team at PT X who have supported the author throughout this research.

REFERENCES

- Aslamiah, S, and Kurniawan, D. (2019) 'Hubungan Pengetahuan dan Sikap Terhadap Kepatuhan Penggunaan Alat Pelindung Diri (APD) di PT. X', Prosiding Seminar Nasional Kesehatan Lingkungan Menghadapi, Kebijakan Ilmu Kesehatan dan Farmasi Universitas Muhammadiyah Kalimantan Timur, pp. 42–45.
- Chung, I., Ryu, H., Yoon, S. Y. and Ha, J. C. (2022) 'Health Effects of Sodium Hypochlorite: Review of Published Case Reports', *Environmental Analysis Health and Toxicology*, 37(1), e2022006. https://doi.org/10.5620/eaht.2022006
- Crini, G. and Lichtfouse, E. (2019) 'Advantages and Disadvantages of Techniques Used for Wastewater Treatment', *Environmental Chemistry Letters*, 17, pp.145–155. https://doi.org/10.1007/s10311-018-0785-9.
- Department of Occupational Safety and Health Ministry of Human Resources Malaysia (2018) A Manual of Recommended Practice on Assessment of The Health Risks Arising From The Use of

Chemicals Hazardous to Health at The Workplace. 3rd edn.

- ECHA (2023) Toxicological Summary Sodium Hydroxide. Available at: https://echa. europa.eu/registration-dossier/-/registereddossier/15566/7/1 (Accessed: 9 January 2024).
- Falakh, F. and Setiani, O. (2018) 'Hazard Identification and Risk Assessment in Water Treatment Plant Considering Environmental Health and Safety Practice', E3S Web of Conferences, ICENIS 2017, 31. https://doi.org/10.1051/ E3SCONF/20183106011
- Franken, R., et al. (2020) 'Ranking of Human Risk Assessment Models for Manufactured Nanomaterials Along the Cooper Stagegate Innovation Funnel Using Stakeholder Criteria', NanoImpact, 17, p.100191. https://doi. org/10.1016/J.IMPACT.2019.100191
- Gitis, V. and Hankins, N. (2018) 'Water Treatment Chemicals: Trends and Challenges', Journal of *Water Process Engineering*, 25, pp.34–38. https:// doi.org/10.1016/j.jwpe.2018.06.003
- Han, D.-H.M.S.Y.-S.C. (2021) 'Regulations on Wearing Personal Protective Equipment by Hazardous Chemical Handlers and Their Implementation', *Journal of Environmental Health Sciences*, 47(1), pp.101–109. https://doi. org/10.5668/JEHS.2021.47.1.101
- ILO (2021) ICSC 0482 Sodium Hypocholrite (Solution, Active Chlorine <10%).
- Kementerian Ketenagakerjaan RI (2022) Profil Keselamatan dan Kesehatan Kerja Nasional Indonesia Tahun 2022. Jakarta.
- Kulkarni, P. and Jeffery, S. (2018) 'The Effects of the use of Diphoterine® Solution on Chemical Burns in the Tarapur Industrial Complex, India', *Burns Open*, 2(2), pp. 104–107. https://doi. org/10.1016/J.BURNSO.2018.03.001
- Manzoor, A.F. (2020) 'Core Competencies of Truck Drivers Responding to Emergencies During Transportation of Hazardous Materials', *Journal* of Health and Pollution, 10(27). https://doi. org10.5696/2156-9614-10.27.200909
- NIOSH (2023) Hierarchy of Controls. Available at: https://www.cdc.gov/niosh/topics/hierarchy/ default.html (Accessed: 21 April 2024).
- Noviarmi, F.S. and Prananya, L.H. (2023) 'Hubungan Masa Kerja, Pengawasan, Kenyamanan APD dengan Perilaku Kepatuhan Penggunaan Alat Pelindung Diri (APD) pada Pekerja Area PA Plant PT X', Jurnal Keselamatan Kesehatan Kerja dan Lingkungan (JK3L), 4(1), pp.57–66. Available at:

http://jk31.fkm.unand.ac.id/index.php/jk31/article/ view/71/59 (Accessed: 6 February 2024).

- Occupational Safety and Health Administration (2018) Dermal Exposure - Overview. Available at: https://www.osha.gov/dermal-exposure (Accessed: 25 January 2024).
- Oppl, R. Kalberlah, F., Evans, P. G. and, & van Hemmen, J. J. (2003) 'A Toolkit for Dermal Risk Assessment and Management: An Overview', *Annals of Occupational Hygiene*, 47(8), pp.629– 640. https://doi.org/10.1093/annhyg/meg069
- Pubchem (2023a) Hydrochloric Acid. Available at: https://pubchem.ncbi.nlm.nih.gov/compound/ Hydrochloric-Acid#section=Hazards-Summary (Accessed: 11 December 2023).
- Pubchem (2023b) Sodium Hydroxide. Available at: https://pubchem.ncbi.nlm.nih.gov/compound/ Sodium-Hydroxide#section=Hazards-Summary (Accessed: 12 December 2023).
- Pubchem (2023c) Sodium Hypochlorite Compound Summary. Available at: https:// pubchem.ncbi.nlm.nih.gov/compound/Sodium-Hypochlorite#section=Hazards-Summary (Accessed: 11 December 2023).
- Pubchem (2024) Sulfuric Acid. Available at: https:// pubchem.ncbi.nlm.nih.gov/compound/Sulfuric-Acid#section=Toxicity-Summary (Accessed: 3 May 2024).
- Riddick, J.M. (2020) 'Hazards of Sodium Hydroxide', Loss Prevention Bulletin, (271), p.13.
- Saputra, A. and Widowati, E. (2023) 'Relationship Between Predisposing Factors and Compliance with the Use of PPE (Personal Protective Equipment) Among Workers at Steel Industry of PT X', *Poltekita: Jurnal Ilmu Kesehatan*, 17(02). https://doi.org/10.33860/jik.v17i2.2135
- Schlueter, U. and Tischer, M. (2020) 'Validity of Tier 1 Modelling Tools and Impacts on Exposure Assessments within REACH Registrations-ETEAM Project, Validation Studies and Consequences', International Journal of Environmental Research and Public Health, 17(12). https://doi.org/10.3390/ijerph17124589
- Slaughter, R.J., Watts, M., Vale, J.A., Grieve, J. R. and Schep, L.J. (2019) 'The Clinical Toxicology of Sodium Hypochlorite', *Clinical Toxicology* (*Philadelphia*, *Pa.*), 57(5), pp. 303–311. https:// doi.org/10.1080/15563650.2018.1543889
- Van Hemmen, J.J. (2005) 'RISKOFDERM: Risk Assessment of Occupational Dermal Exposure to Chemicals', Journal for Applied Occupational Health and Safety Science, 1.